

Note

**Reaction of the D-ribose moiety of adenosine
and AMP with periodate and 5,5-dimethylcyclohexane-
1,3-dione (dimedone)**

FRITZ HANSSKE AND FRIEDRICH CRAMER

*Max-Planck-Institut für experimentelle Medizin, Abteilung Chemie,
34 Göttingen, Hermann-Rein-Straße 3 (Germany)*

(Received July 18th, 1974, accepted for publication in revised form, November 25th, 1974)

The *cis*-glycol group of ribonucleosides and ribonucleotides can be oxidized specifically by periodate to form dialdehydes (**1** → **2**) which subsequently polymerise to oligo- or poly-ribo-oxynucleosides and -ribo-oxynucleotides¹

On investigating the reaction of the dialdehyde **2** (R = H), derived from adenosine, with dimedone, a single product was isolated, the combustion analysis of which was consistent with structure **4** (R = H). The mass spectrum of **4** (R = H) exhibits a molecular ion at *m/e* 527 and therefore provides further evidence for the presence of two dimedone residues. Furthermore, in the n m r spectrum of **4** (R = H), the two acidic protons (H-2'') of the dimedone residues could be identified by D₂O exchange. The splitting of the signals for the methyl and methylene protons of the

TABLE I

ELECTROPHORESIS, CHROMATOGRAPHY, AND U V SPECTROSCOPY OF **2-4**

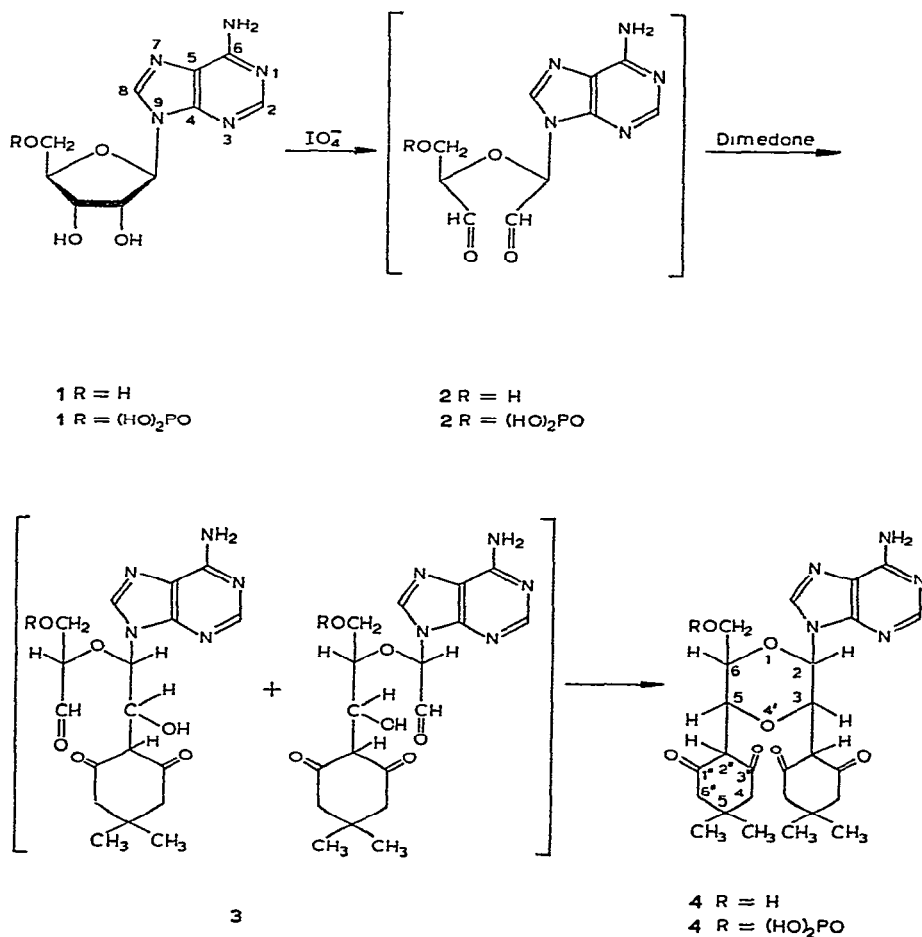
Compound	Electrophoretic mobilities ^a		<i>T l c</i> R _F ^b	<i>U v spectra</i>	
	A	B		λ_{\max} (nm)	$\lambda_{\text{shoulder}}$ (nm) ^c
2 R = (HO) ₂ PO	0.37 (+)	1.0 (+)	0.85	258	—
2 R = H	—	0.23 (—)	0.74	258	—
Dimedone	0.59 (+)	0.7 (+)	0.55	280	—
3 R = (HO) ₂ PO	0.28 (+)	—	0.60	265	286 (weak)
4 R = (HO) ₂ PO	1.0 (+)	—	0.22	264.5	290
3 R = H	—	0.0	0.30	265	286 (weak)
4 R = H	0.27 (—)	0.24 (+)	0.15	264.5	290

^aA, Cellulose thin-layer plates (Merck, Darmstadt, GFR), 0.1M acetate buffer (pH 5.0), 600 V, 31 → 51 mamp; B, silica gel thin-layer plates (Woelm, Eschwege, GFR), 0.1M Sørensen citrate buffer (pH 6.5), 600 V, 29 mamp (apparatus from Desaga, Heidelberg, GFR). ^bSilica gel (as above), 0.25M LiCl. ^c0.1M Phosphate buffer (pH 7.0).

dimedone moieties may be based on interactions with the purine ring, and on the fact that C-3' and C-5' are additional asymmetric centers and that, therefore, stereoisomers can be formed. From these results, together with chromatography, electrophoresis, and u v data (Table I), we suggest the structure 4. From u v and electrophoresis data, 3 or structures generated from 3 by ring closure are proposed as intermediates.

The reaction product of 2 [$R = (HO)_2PO$], derived from adenosine monophosphate (AMP), and dimedone is probably 4 [$R = (HO)_2PO$], since dephosphorylation by alkaline phosphatase yields 4 ($R = H$).

Glitz and Sigman² treated the dialdehyde 2, derived from AMP, as well as several periodate-oxidized RNA species, with [^{14}C]-labelled dimedone under slightly different conditions. From the incorporation of radioactivity and u v data, they concluded that a 1:1 stoichiometry of the dialdehyde to dimedone existed in these compounds.



EXPERIMENTAL

Adenosine, AMP, and dimedone were commercial products of analytical grade. Alkaline phosphatase from *E. coli* (E C 3131) was a product of Boehringer, Mannheim. N m r. spectra were measured at 24° on a Varian HA-100 spectrometer, using methyl sulfoxide-*d*₆ as solvent and tetramethylsilane as internal standard. Mass spectra were obtained on a Varian-MAT CH-4 spectrometer. I r spectra were measured in KBr pellets, using a Perkin-Elmer Infracord spectrometer. Zeiss PMQ II and UNICAM SP 1800 spectrometers were used for measuring the u v spectra.

Preparation and characterization of 6-amino-9-[3',5'-bis(5'',5''-dimethylcyclohexane-1'',3''-dione-2''-yl)-6'-hydroxymethyl-1',4'-dioxan-2'-yl]purine (4, R = H) — Adenosine (10 mmol) was treated with 10 mmol of NaIO₄ in 100 ml of water at 23° in the dark for 1 h. Traces of periodate then were reduced³ with butane-2,3-diol. A solution of dimedone (20 mmol) in 600 ml of water was filtered and added to the foregoing solution, and the mixture was stored at 23° overnight. The product was collected, recrystallized from aqueous methanol, and dried at 80° over P₂O₅ *in vacuo* to give **4** (4.5 g, 81%; R = H), m p 184.5° (Found: C, 57.4, H, 6.56, N, 12.97. C₂₆H₃₃N₅O₇ · H₂O calc: C, 57.3, H, 6.46, N, 12.83%).

On boiling a solution of **4** (R = H) in ethanol, the anhydrous compound was obtained, m p beginning at 190° (Found: C, 59.02, H, 6.50, N, 13.45. C₂₆H₃₃N₅O₇ calc: C, 59.1; H, 6.32, N, 13.29%).

At pH 3, **4** (R = H) has λ_{max} 260 nm (ε 25,800), pH 7, λ_{max} 264 nm (ε 24,500), shoulder at 290 nm, pH 10.5, λ_{max} 265 nm (ε 26,500), shoulder at 290 nm, ν_{max} 1715 (weak, ketone of a six-membered ring) and 1620 cm⁻¹ (strong, 1,3-diketone in the enolic form). Mass spectrum *m/e* 527 (6.5%) M⁺ - H₂O, 509 (21.5) 527 - H₂O, 491 (20) 509 - H₂O, 473 (9) 491 - H₂O, 136 (100) BH⁺, 135 (100) B⁺, 392 (13) 527 - B, 374 (95) 509 - B, 356 (93) 481 - B, 338 (92) 473 - B, 479 (53) 509 - CH₂O, 344 (73) 479 - B. N m r. data: δ 8.2 (*s*, 1 H, H-8), 8.18 (*s*, 1 H, H-2), 7.23 (*s*, 2 H, NH₂), 6.76 (*s*, 1 H, H-2'), 4.18 (*d*, *J* 4.5 Hz, after D₂O exchange *s*, 1 H, H-3'), 3.5 (*s*, 2 H, H-5',6'), 3.34 (*s*, 2 H, CH₂-6'), 1.03 (*s*) and 0.87 (*d*) (12 H, 4 Me-5''), 2.3 (*m*, 8 H, 4 CH₂-4'',6''), 5.75 (*d*, *J* 4.5 Hz, after D₂O exchange *s*, 2 H, CH-acid-2''). Sometimes, the sugar protons exhibited coupling constants of 1–2 Hz, but in most cases, no splitting of the signals could be detected.

Synthesis of 6-amino-9-[3',5'-bis(5'',5''-dimethylcyclohexane-1'',3''-dione-2''-yl)-6'-phosphoryloxymethyl-1',4'-dioxan-2'-yl]purine [4, R = (HO)₂PO] and further preparations of **4** (R = H) were carried out in 10 mM solution by treating **1** with 1 equiv of NaIO₄ for 30 min at 4° in the dark and subsequently with 2 equiv of dimedone in 15 mM Teorell-Stenhagen buffer, either at pH 4.9 or 6.9 or 8.8, for 12 h at 4°. Analyses were carried out as described above.

Ester cleavage with alkaline phosphatase — A reaction mixture (50 μl) of 0.5 μmol of **2** [R = (HO)₂PO] and 1 μmol of dimedone were incubated with 10 μl of alkaline phosphatase solution (= 1 μg of enzyme) at pH 8.8 and 37° for 40 min. The reaction products were worked up and analyzed as described above.

REFERENCES

- 1 F HANSKE, M SPRINZL, AND F CRAMER, *Bioorg Chem*, 3 (1974) 367.
- 2 D G GLITZ AND D S SIGMAN, *Biochemistry*, 9 (1970) 3433
- 3 M L WOLFROM AND J M BOBBITT, *J Amer Chem Soc*, 78 (1956) 2489